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Name Date Class 10-1 Review and Reinforcement (continued)

Find the formula mass of each of the following compounds. Show all your work. 11. CaCO_3 PDF www.ranchorams.org '16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established.

Chemistry Chapter 10 Review And Reinforcement

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Answers

10-1 Review and Reinforcement (continued) Find the formula mass of each of the following compounds. Show all your work.

11. CaCO_3 12. MgSO_4 13. NaOH 14. KCl 15. BaF_2 o JC 04) 137B
Find the molar mass of each of the following compounds. Show all your work. 16. nitric acid (HNO_3) (D 17. ammonia (NH_3) 18. calcium sulfate (CaSO_4) 19. potassium sulfide (K_2S)

Name Date Class 10-1 Review and Reinforcement (continued ...

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10-1 Review and Reinforcement (continued)
Find the formula mass of each of the following compounds. Show all your work.
11. CaCO_3 12. MgSO_4 13. NaOH 14. KCl 15. BaF_2 o
JC 04) 137B Find the molar mass of each of the ...

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11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. condensation of matter coefficients subscripts mass

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1. from the given formula, you take the amount of each element, divide by its atomic mass then times 100 2. experimental analysis. a sample of an unknown compound with a mass of 0.2370g is extracted from the roots of a plant. decomposition of

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the sample produces 0.09480 g of C, 0.1264 g of O, and 0.0158 g of H. what is the percent composition ...

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10—2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q \$ gK2S 10. 65.8gMg= l. atoms Mg Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number

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of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO₂ gas.

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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of a liquid placed in. a container, equilibrium is established when the Equid vaporizes at the same rate that vapor condenses. 7, NO, in reversible

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reactions at equilibrium,

Rancho High School

Read PDF 11 3 Review And Reinforcement Answers 3—1 Review and Reinforcement Early Models of the Atom On the line at the left, write the letter of the chemist who proposed each of the ideas listed below. Each letter may be used once, more than once, or not at all. a. Democritus b. Lavoisier c. Proust d. Dalton 1.

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10-4 Review and Reinforcement Chemical Measurements Define each of the following terms in complete sentences. 1. mole 2. formula mass 3. formula unit 4. Avogadro's number 5. molar mass If the statement is true, write "true." If it is false, change the underlined word or words to make it true. _____ 6.

Name Date Class of following terms in complete sentences.

10-1 Review and Reinforcement (continued) Find the formula mass of each of the following compounds Show all your work 11 CaCO_3 12 MgSO_4 13 NaOH 14 KCl 15 BaF_2 (JC 04) 137B Find the molar mass of each of the following compounds Show all

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