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5. Cs-137 ($t_{1/2}=30.2$ -years) is produced as a waste product in nuclear fission reactors. What fraction (N/N_0) remains undecayed after 250.0-years? 10. If your cellar was measured to contain 2.4-g of Rn-222 ($t_{1/2}=3.8$ -days) (a radioactive gas naturally produced by some granite deposits), how long would it take for that sample to decay to 0.2-g?

5-2 Review and Reinforcement - Steffin

5-2 Review and Reinforcement, Reading the Periodic Table . On
Page 2/11

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the line at the left, write the letter of the appropriate location of each group of elements on the periodic table below. Some letters will be used more than once. 1. carbon family 8. f-block elements 2. alkaline earth metals 9. noble gases 3. inner transition metals 10. p-block ...

5-2 Review and Reinforcement - MRS. Nairn

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2-5 Review and Reinforcement - Weebly 5.2 REINFORCEMENT
5.2.1 Description This work shall consist of furnishing and placing bars of the grade, type and size shown in accordance with these Specifications and in conformity with the

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6—2 Review and Reinforcement (continued) Identify each of the following elements as a metal (M), nonmetal (NM), or semimetal (SM). 19. sodium —SM— silicon 21. neon 22. 23. nitrogen Write the family names that have been given to each of the following groups. 24. Group IA alkali 25. Group 2A 26. Group 7 A 27.

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Group8A hvohlt

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CAEn0 .1 //omEL0đRK S 8-1 Review and Reinforcement, pages 12—13 8—2 Review and Reinforcement, pages 19—20 polar I polar 1. 2. 3. 5. 6. 8.

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2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the dry cell battery. The chemical energy in the

2 Answer - Weebly

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16—3 Review and Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperature affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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2-5 Review and Reinforcement . Mixtures . Use what you have learned in Section . 2-5 . to identify each of the following substances as pure substances, heterogeneous mixtures, or homogeneous mixtures. Write "pure," "heterogeneous," or "homogeneous" on the line. 1. alphabet soup 2. salt 3. concrete

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4. vegetable oil 5. air 6. paint . 7. sea water 8. granite 9. steel

2-5 Review and Reinforcement - MRS. Nairn

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11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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